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Solid-Phase Synthesis of New S-Glycoamino Acid Building Blocks

Laurence Jobron and Gerd Hummel*

Jerini Bio Tools GmbH, Rudower Chaussee 29, 12489 Berlin, Germany hummel@jerini.de

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ABSTRACT

Efficient synthesis of unprotected S-glycoamino acid building blocks in the solid phase by coupling a sugar 1-thiolate with iodine activated fluoren-9-ylmethoxycarbonyl (Fmoc) protected amino acids.

Glycoconjugates have been implicated in many biological events important in inflammation, immune response, and tumor metastasis.¹ Of special interest are glycoproteins containing modified glycosyl amino acids, thus exhibiting new properties. While *S*-glycopeptides have been isolated from nature,² the driving force behind the synthesis of *S*-glycosides has been the production of glycopeptidomimetics with enhanced stability toward chemical and enzymatic degradation.³ Several attempts have been made to introduce the *S*-linkage by chemical synthesis.⁴ A variety of glycosylation methods have been applied including Koenigs—Knorr,⁵ glycosyl fluorides,⁶ Lewis acid-catalyzed glycosylation,⁷ trichloroacetimidates⁸ and isothiouronium salts.⁹ These methods generally use protected carbohydrates and cysteine derivatives. We present here the solid-phase

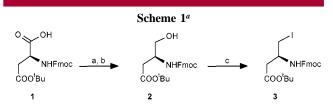
key feature of this method is that a nucleophilic sugar 1-thiolate without protective groups is used for coupling with an iodine activated fluoren-9-ylmethoxycarbonyl (Fmoc)/t-Bu protected amino acid.

Hummel and Hindsgaul¹⁰ described the use of a sugar-1-

synthesis of new S-glycoamino acid building blocks. The

Hummel and Hindsgaul¹⁰ described the use of a sugar-1-thiolate without protective groups in the solid phase as the nucleophile for coupling with trifluoromethanesulfonate (triflate)-activated glycosides. Since the use of triflates with Fmoc-protected amino acids is not compatible, we decided to prepare amino acid iodo derivatives for the synthesis of *S*-glycoamino acid building blocks.

First the free acid functions of the *N*-Fmoc/*t*-Bu ester protected glutamic and aspartic acid derivatives were reduced to the corresponding alcohols **2**, **4**, **6**, and **8** (Scheme 1,



 $^{\rm a}$ Reagents and conditions:(a) CICO $_2$ C $_2$ H $_5$, Et $_3$ N, THF, -30°C, 30 min; (b) NaBH $_4$, H $_2$ O, THF, 0°C to 20°C, 4 h, 74%; (c) I $_2$, PPh $_3$, imidazole, PhCH $_3$, 120°C, 20 min, 80% .

⁽¹⁾ Varki, A. *Glycobiology* **1993**, *3*, 97. Lee, Y. C.; Lee, R. T. *Acc. Chem. Res.* **1995**, *28*, 322. Chambers, W. H.; Brisette-Storkus, C. S. *Chem. Biol.* **1995**, *2*, 429.

⁽²⁾ Lote, C. J.; Weiss, J. B. *Biochem. J.* **1971**, *123*, 25p. Lote, C. J.; Weiss, J. B. *FEBS Lett.* **1971**, *16*, 81. Weiss, J. B.; Lote, C. J.; Bobinski, H. *Nature New Biol.* **1971**, *234*, 25.

⁽³⁾ Michael, K.; Wittmann, V.; König, W.; Sandow, J.; Kessler, H. Int. J. Pept. Protein Res. 1996, 48, 59.

⁽⁴⁾ Taylor, C. M. Tetrahedron 1998, 54, 11317.

⁽⁵⁾ Baran, E.; Drabarek S. Pol. J. Chem. 1978, 52, 941. Gerz, M.; Matter, H.; Kessler, H. Angew. Chem., Int. Ed. Engl. 1993, 32, 269.

⁽⁶⁾ Nicolaou, K. C.; Chucholowski, A.; Dolle, R. E.; Randall, J. L. J. Chem. Soc., Chem. Commun. 1984, 1155.

⁽⁷⁾ Salvador, L. A.; Elofsson, M.; Kilberg J. Tetrahedron 1995, 51, 5643.

⁽⁸⁾ Käsbeck, L.; Kessler, H. Liebigs Ann./Recueil 1997, 165.

⁽⁹⁾ Monsigny M. L. P.; Delay, D.; Vaculik, M. Carbohydr. Res. 1977, 59, 589.

⁽¹⁰⁾ Hummel, G.; Hindsgaul, O. Angew. Chem., Int. Ed. 1999, 38, 1782.

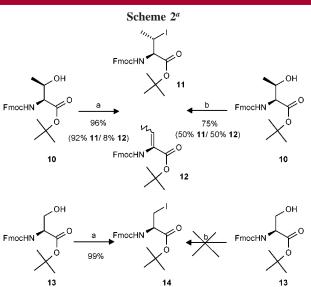
shown for Fmoc-Asp(O'Bu)) using ethyl chloroformate and sodium borohydride in tetrahydrofuran (yields are given in Table 1).

Table 1

amino acids	reduction to alcohol	conversion into iodide
Fmoc-Asp(O ^t Bu)-OH	2 (74%)	3 (80%)
Fmoc-Asp-O'Bu	4 (81%)	5 (81%)
Fmoc-Glu(O'Bu)-OH	6 (86%)	7 (95%)
Fmoc-Glu-O'Bu	8 (83%)	9 (76%)

Treatment of the different alcohols **2**, **4**, **6**, and **8** with triphenylphosphine—iodine—imidazole¹¹ in toluene at 120 °C afforded the corresponding iodo derivatives **3**, **5**, **7**, and **9** in 76–95% yields after purification by flash chromatography on silica gel.

The same procedure applied to Fmoc-Thr-O'Bu 10 and Fmoc-Ser-O'Bu 13¹² gave only low yields with threonine and no product with serine (Scheme 2). In the case of



^aReagents and conditions: (a) PDPI, imidazole, CH_2CI_2 , reflux; (b) I_2 , PPh₃, imidazole, PhCH₃, 120°C, 20 min

threonine, we observed the formation of elimination product **12** and iodo compound **11** in a ratio of 1:1. No better results could be obtained by decreasing reaction time or temperature; therefore we tried different conditions. The use of polystyryl diphenylphosphine—iodine (PDPI) complex was previously described¹³ for the conversion of *N*-Fmoc β -amino alcohols into their corresponding iodides in high yields and without any detectable epimerization of the chiral center. Similar

reaction of the alcohols 10 and 13 with triphenylphosphine—iodine complex in the presence of imidazole in anhydrous dichloromethane gave the iodides 11 (with inversion of configuration) and 14 in high yield (11 88% and 14 99%) under very mild conditions (3 h reflux for Thr 11 and 10 min for Ser 14). No elimination product was observed with Ser and only 7% in the case of Thr. Both iodo compounds were used without further purification in the following steps.

Thiol **15** immobilized on a trityl chloride derivatized polystyrene resin was obtained after reduction of the disulfide¹⁰ (Scheme 3). The loading of the thiol on the solid

*Reagents and conditions: (a) NaOMe, THF, 20°C, 2 h; (b) [15] Crown-5, iodo derivatives **3-6**, THF, 45°C, 3 d; (c) 50% TFA, CH₂Cl₂, 20°C, 30 min.

support was determined by elemental analysis (sulfur content) and was 1.2 mmol/g. **15** was treated for 2 h with NaOMe/THF to enhance its nucleophilicity¹⁰ and washed with a mixture of MeOH/THF to eliminate traces of NaOMe. The resulting sodium thiolate was coupled with amino acid iodo derivatives **3**, **5**, **7**, **9**, and **14** in THF at 45 °C in the presence of [15]crown-5 as complexing agent. After 3 days, the resin was washed and the glyco amino acids were cleaved from the resin by treatment with 50% TFA in CH₂Cl₂. All final derivatives **16**–**20** were obtained after reverse-phase HPLC purification in good yields (75 to 82%).

Reaction of the immobilized thiolate with threonine derivative 11 however gave only poor yields (<5%).

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⁽¹¹⁾ Garegg, P. J.; Samuelsson, B. J. Chem. Soc., Chem. Commun. 1979

⁽¹²⁾ Liebe, B.; Kunz, H.. Angew. Chem., Int. Ed. Engl. 1997, 36, 618.

⁽¹³⁾ Caputo, R.; Cassano E.; Longobardo, L.; Palumbo, G. $Tetrahedron~\mathbf{1995},~51,~12337.$

To achieve better results we studied the influence of different conditions such as solvent (THF or DMF), temperature (20 or 45 °C), and complexing agent ([15]crown-5 or Kryptofix 221). The results are summarized in Table 2.

Table 2

temperature	solvent	complexing agent	yield, %
20 °C	THF	[15]crown-5	<20
		Kryptofix 221	
20 °C	DMF	[15]crown-5	30
		Kryptofix 221	
45 °C	THF	[15]crown-5	< 5
		Kryptofix 221	
45 °C	DMF	[15]crown-5	50
		Kryptofix 221	<10

Best yields were obtained using DMF at 45 °C in the presence of [15]crown-5 as complexing reagent (Scheme 4). We observe that compound **21** reverts back to its threo configuration after the substitution.

In conclusion, we present here an efficient synthesis of new S-glycoamino acid building blocks in the solid phase which are ideal building blocks for the solid-phase synthesis

Scheme 4^a

*Reagents and conditions: (a) NaOMe, THF, 20°C, 2 h; (b) [15] Crown-5, 11, DMF, 45°C; (c) 50% TFA, CH₂Cl₂, 20°C, 30 min.

of *S*-glycopeptides using the Fmoc strategy. All glycosides were obtained stereoselectively and in high yields. Using iodine-activated amino acids gives the corresponding *S*-glycosides under very mild conditions with retention of configuration (double inversion) if the iodine is part of an asymmetric carbon atom.

Supporting Information Available: Experimental procedures for preparation of compounds 2–9, 11, 14, and 16–21 and analytical data for 2–9, 11, 14, and 16–21. This material is available free of charge via the Internet at http://pubs.acs.org.

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